# Electronic Structure of Ge:GeO<sub>2</sub> interfaces for future CMOS

John Robertson, H Li, L Lin, K Xiong\*, Engineering Dept, Cambridge University, UK \*UT Dallas, jr@eng.cam.ac.uk

- Need for Ge
- Problems of Ge
- Defects in GeO<sub>2</sub>
- Hydrogen in GeO<sub>2</sub>

# **Future CMOS**



- Replace SiO2 with high K oxide, HfO2
- Replace poly-Si gate with metal gate
- Replace Si channel with high mobility channel
- Change geometry

# **Need for Ge**

- Ge has higher carrier mobilities than Si, particularly holes
- Ge pFET, GaAs nFET
- But Ge nFET, pFET also possible

	Si	Ge	GaAs
Band Gap (eV)	1.1	0.66	1.42
Electron mobility (cm <sup>2</sup> /V.s)	1500	3900	8500
Hole mobility (cm <sup>2</sup> /V.s)	450	1900	400

# **Problems**

- GeO<sub>2</sub> Lack of insulating properties
- GeO<sub>2</sub> poor passivation
- GeO volatilisation
- Fermi level pinning near VB
  for nFET

# **Thermodynamics etc**

- GeO<sub>2</sub> is considerably less stable than SiO<sub>2</sub>
- Ge<sup>2+</sup> more stable
- Band gap much less
- Ge-H bond strength not much less

	SiO <sub>2</sub>	GeO <sub>2</sub>
$\Delta H_{f}$ (eV)	-4.80	-3.27
Band gap (eV)	9.0	6.0
Si-H bond (eV)	3.3	3.1
CNL (of semiconductor) (eV)	0.3	0

## **Ge:GeO<sub>2</sub> interface**

 Calculated 4.3 eV VB offset





CUED

## GeO<sub>2</sub> Band Gap, Band Offsets from Photoemission

- A Ohta, S Miyazaki et al, eJ Surf Sci Nanotech 4 174 (2006)
- Band gap

#### **VB** offset



#### Band gap, band offsets

9 Band gap is much 8 lower for GeO<sub>2</sub> due to smaller CB offset 7 6 0.8 eV 3.3 eV • VB offset almost 5 Energy (eV) unchanged (O-like 4 character of VB top) 6.1 eV 9 eV 3 4.3 eV 4.5 eV 2 Calculation using Screened Exchange 0 Ge GeO2 SiO2

# **GeO**<sub>2</sub> as gate oxide

 Small CBO explains use of relatively thick GeO2 layer in Ge FETs with GeO2 gate oxide (Toriumi, IEDM 2009)

# **GeO volatilisation**

- GeO evolution causes defects and worse electrical behavior
- Needs supply of Ge to occur –
  - not for GeO<sub>2</sub> on Si
- GeO desorbs from surface
- O vac diffusion through GeO<sub>2</sub>
- Toriumi (JJAP 2008), Kita (IEDM 2009, JAP 2010)



# Si oxidation

- During initial stages of Si oxidation (< 5nm) reactive layer model
- Deal-Grove model based on O2 diffusion (interstitial)
- O<sup>17</sup> isotope tracer analysis suggests not O<sub>2</sub> diffusion (Rochet, Adv Phys 35 237 1986)
- But Baumvol, PRB 60 1492 (1999) shows no mobility of Si<sup>29</sup>.
- Hence O diffusion

## **Solid GeO structure**

- GeO molecule
- Geo solid, at interfaces
- What is its structure?
- Iso-electronic to PbO
- But has structure of GeS, but with planar O site (as in Si3N4)
- Lin, Robertson, APL (2010)



PbO GeS GeO

## **GeO states**

- GeO has filled Ge s states,
- VB of Ge s, O 2p.
- CB of Ge p states



## **Ge:GeO interface**

 Ge:GeO epitaxial model used to calculate band offsets





#### CUED

# **Ge:GeO**<sub>2</sub> interface

- Presence of GeO<sub>2</sub> and GeO at interface with small CBO is a trap
- means that GeO<sub>2</sub> should be avoided
- But GeO<sub>2</sub> needed to stop mobility degradation
- To stop poor reliability



# **Defects in SiO**<sub>2</sub>

- E' centers
- Neutral Oxygen vacancy in SiO<sub>2</sub> relaxes to a Si-Si bond





# **Defects in GeO**<sub>2</sub>

- E' centers
- Neutral Oxygen vacancy in GeO<sub>2</sub> relaxes to a Ge-Ge bond
- No states in gap (GGA)





#### **Wavefunctions**



#### CUED

# **Novel defects in GeO<sub>2</sub>**

- Ge-Ge bond breaks
- One 3-fold Ge atom flips through Ge-O, to bond to back Oxygen
- Makes 3-fold Ge +3-fold
  O



# **Novel defects in GeO<sub>2</sub>**

- 3-fold G gives gap state (sX)
- 3-fold O gives state at CB edge, localised on adjacent Ge sites
- Similar to 'Valence alternation pairs in GeO<sub>2</sub>' by Pasquarello et al, APL (2010)



#### wavefunctions



#### CUED

#### **Poor interface**

- Si:SiO<sub>2</sub> interface is abrupt and smooth for T<1100C.</li>
- Low scattering
- SiO<sub>x</sub> dissociates into Si and SiO<sub>2</sub> (Lucovsky, JNCS 227 1 (1998)
- GeOx would not do same



## solutions

- Remove GeO<sub>2</sub> layers
- LaGeO<sub>x</sub> (Dimoulas etc)
- GeSr, etc (Kamata)



# **Improving mobility in HK-MG**

- Separate HfO<sub>2</sub> from channel by 1 nm of SiO<sub>2</sub> improves mobility by screening remote scattering
- K Maitra,..IBM, JAP (2007)



CUED

# **Role of SiO<sub>2</sub> interfacial layer**

- SiO<sub>2</sub> interfacial layer is retained,
- To limit mobility degradation due to remote phonon scattering, remote Coulomb scattering
- To limit interfacial defects/ reliability problems





# **Mobility degradation**

 GeO<sub>2</sub> interlayer may also be needed to lessen degradation



# **Defect passivation**

- Interface defects such as P<sub>b</sub> centre (Ge dangling bond)
- Why is defect density D<sub>it</sub> high?
- Why does not Hydrogen passivate them?

#### **Ge dangling bonds**

- Baldovino, APL 93 242105 (2008) does find Ge DB by electrically detected ESR
- Stesmans finds no ESR for 100% Ge (PRB 2009)



# **Ge dangling bonds**

- Houssa says GeO<sub>2</sub> relaxation removes DBs
- Does not explain why so many Dit's !



#### **Ge dangling bond + Hydrogen**

- Poor passivation of Ge P<sub>b</sub> centres attributed to
- Ge dangling bond lying below VB edge (Janotti + van de Walle, APL 2007)
- Pasquarello found it ok
- That is Ge<sup>-</sup> and H<sup>-</sup> repel



#### Hydrogen in Ge

 H c\an diffuse through Ge





# H in GeO2

- H2 can diffuse through GeO2 as H2 interstitial
- And react with Ge dangling bond



#### **Interstitial H atom**



CUED

# H in GeO<sub>2</sub>

- Unlike in SiO2, H in GeO2 is a donor, level likes just at CB edge
- Donates electron
- Donates electron to Ge

8

6

2

0

Energy (eV) 5



#### H moves from Ge-H to O site

- H may go to O bond, not to Ge DB
- Tsertis and Pantelides, APL (2010)



## Conclusions

- Ge is not so like Si
- Poor band offset avoid GeO<sub>2</sub> interfacial layer
- Non-stoichiometry
- Role of hydrogen